Development of Calibration-Free Autonomous Chemical Sensors for Ocean Biogeochemical Measurements

Michael DeGrandpre Department of Chemistry The University of Montana Missoula, MT 59812

phone: (406) 243-4118 fax: (406) 243-4227 email: mdegrand@selway.umt.edu

Award #: N00014-00-1-0573 http://www.onr.navy.mil/sci_tech/ocean/onrpgahj.htm http://www.cas.umt.edu/chem/degrandp/

LONG-TERM GOALS

The long term goal of the project is to develop autonomous chemical sensors for oceanographic applications. The autonomous sensors will be used on moorings, drifters and other unmanned platforms to study ocean biogeochemical processes. The results obtained from the autonomous sensors will help quantify ocean biogeochemical fluxes and will be used to further develop coupled physical-biogeochemical models of the surface ocean. Improvements in current models will allow more accurate predictions of the effects of climate change and other anthropogenic impacts on ocean biogeochemical cycles.

OBJECTIVES

The primary objective is to determine if other chemical sensors can be developed based on the calibration-free design discovered by the PI during the development of a pCO_2 sensor (DeGrandpre et al. 1995, 1999). If the design proves applicable to other analytes, it will be possible to develop autonomous sensors for these analytes that have an identical sensor-to-sensor response with excellent long-term stability, potentially eliminating the need for periodic calibrations.

APPROACH

The analytes total ammonia, silicate, and calcium will be initially investigated. The method requires that an indicator chemistry be used that has two different colored forms in the presence of the analyte as described for the pCO_2 sensor in DeGrandpre et al. (1999). The absorbances of the complexed and uncomplexed forms of the colored reagent (indicator) are combined to obtain a response that is only dependent upon the indicator equilibrium composition. Different reversible reagent chemistries that have been reported in the literature will be investigated. We have proposed to utilize an acid-base indicator for detection of total ammonia. We found a possible alternative to the standard molydenum blue reaction for silicate, which utilizes the first step of the standard method, formation of a reversible molybdosilicic acid complex. Others have improved the sensitivity by formation of a reversible ion associate with the complex (Saurina and Hernandez-Cassou 1995). We plan to design the Ca sensor after that reported by Chau and Porter (1990) which utilized calcichrome and an ion exchange membrane. Calcichrome has different absorption spectra in the calcium-complex and uncomplexed forms making it an ideal indicator for testing the calibration-free methodology.

Public reporting burden for the col maintaining the data needed, and c including suggestions for reducing VA 22202-4302. Respondents shot does not display a currently valid C	ompleting and reviewing the collect this burden, to Washington Headqu ald be aware that notwithstanding an	tion of information. Send commentarters Services, Directorate for Inf	s regarding this burden estimate formation Operations and Reports	or any other aspect of the s, 1215 Jefferson Davis	his collection of information, Highway, Suite 1204, Arlington
1. REPORT DATE SEP 2000		2. REPORT TYPE		3. DATES COVE 00-00-2000	ered O to 00-00-2000
4. TITLE AND SUBTITLE		5a. CONTRACT NUMBER			
Development of Ca	Sensors for	5b. GRANT NUMBER			
Ocean Biogeochemical Measurements				5c. PROGRAM ELEMENT NUMBER	
6. AUTHOR(S)				5d. PROJECT NUMBER	
				5e. TASK NUMBER	
				5f. WORK UNIT NUMBER	
7. PERFORMING ORGANIZATION NAME(S) AND ADDRESS(ES) Department of Chemistry, The University of Montana,, Missoula,, MT, 59812				8. PERFORMING ORGANIZATION REPORT NUMBER	
9. SPONSORING/MONITORING AGENCY NAME(S) AND ADDRESS(ES)				10. SPONSOR/MONITOR'S ACRONYM(S)	
				11. SPONSOR/MONITOR'S REPORT NUMBER(S)	
12. DISTRIBUTION/AVAIL Approved for publ		ion unlimited			
13. SUPPLEMENTARY NO	TES				
14. ABSTRACT					
15. SUBJECT TERMS					
16. SECURITY CLASSIFIC	ATION OF:		17. LIMITATION OF ABSTRACT	18. NUMBER OF PAGES	19a. NAME OF RESPONSIBLE PERSON
a. REPORT unclassified	b. ABSTRACT unclassified	c. THIS PAGE unclassified	Same as Report (SAR)	4	RESI ONSIDEL I ERSON

Report Documentation Page

Form Approved OMB No. 0704-0188

WORK COMPLETED

We have begun a thorough literature search to determine if the above reagents are the best possible reagents for each analyte. Equipment has been purchased including a new research grade UV-VIS spectrophotometer that was included as part of the grant. A graduate student working on the project has designed a membrane gas exchange cell for measurement of total ammonia. Plastic fabrication components have been purchased for the flow cell. We are also evaluating new optical design for the sensing systems including a completely integrated fiber optic filter spectrometer and other miniature spectrographs. I am also currently recruiting additional graduate students to work on the project.

RESULTS

Because we are in the early stages of the funding there are no results to date.

IMPACT/APPLICATION

There is a great demand for low maintenance chemical analyzers and sensors that provide high quality data. If the technology used for the PI's pCO_2 sensor proves applicable to other analytes, new sensors would become available for a wide-array of applications. Although our interest lies in the area of ocean biogeochemistry, autonomous chemical sensors would also find wide use in industrial, biomedical, environmental, and defense-related applications.

TRANSITIONS

Our work has been published in the peer-reviewed literature and others may be pursuing development of their own calibration-free designs based on these papers. The pCO_2 sensors are now commercially available (see sunburstsensors.com) and we plan to continue to spin-off new sensor technology to the private sector to make the sensors available to researchers, government agencies, etc.

RELATED PROJECTS

At this point there are no related projects being pursued.

REFERENCES

Chau, L.-K. and M.D. Porter. (1990). Optical sensor for calcium: performance, structure, and reactivity of calcichrome immobilized at an anionic polymer film. *Anal. Chem.*, 62, 1964-1971.

DeGrandpre, M.D., Hammar, T.R., Smith, S.P., and F.L. Sayles. (1995). In situ measurements of seawater *p*CO₂. *Limnol. Oceanog.*, 40, 969-975.

DeGrandpre, M.D., Hammar, T.R., Wallace, D.W.R., and C.D. Wirick. (1997). Simultaneous mooring-based measurements of seawater CO₂ and O₂ off Cape Hatteras, North Carolina, *Limnol. Oceanog.*, 42, 21-28.

DeGrandpre, M.D., Hammar, T.R. and C.D. Wirick. (1998). Short term pCO_2 and O_2 dynamics in California coastal waters. *Deep-Sea Res. Part II*, 45, 1557-1575.

DeGrandpre, M.D., Baehr, M.M. and T.R. Hammar. (1999). Calibration-free optical chemical sensors, *Anal. Chem.*, 71, 1152-1159.

Saurina, J. and S. Hernandez-Cassou. (1995). Flow injection spectrophotometric determination of silicate based on the formation of an ion associate between molybdosilicate and malachite green. *Analyst*, 120, 2601-2604.

PUBLICATIONS

None